

COMBO - ଓଡ଼ିଶା
32,000- PYQ
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61. The most electropositive halogen is
(A) F
(B) Cl
(C) Br
(D) I
62. Elements having same valence electrons are placed in same group but there is an exception with
(A) Ne (B) He
(C) Ar (D) Xe
63. Which of the following gases will have the highest rate of diffusion ?
(A) O₂ (B) CO₂
(C) NH₃ (D) N₂
64. The potential energy of a molecule on the surface of liquid compared to one inside the liquid is
(A) Zero
(B) Smaller
(C) The same
(D) Greater
65. On increasing the concentration of reactants in a reversible reaction, the equilibrium constant will
(A) Depend on the concentration
(B) Increase
(C) Unchanged
(D) Decrease
66. Bee sting leaves an acid which causes pain and irritation. The acid released is
(A) Tartaric acid
(B) Citric acid
(C) Ethanoic acid
(D) Methanoic acid
67. What is called arenes ?
(A) Aromatic hydrocarbons
(B) Alkynes
(C) Unsaturated hydrocarbons
(D) Saturated hydrocarbons
68. Which of the following do not contain carbon-oxygen double bonds ?
(A) Ketone (B) Esters
(C) Acids (D) Ethers
69. Which among the following is the mass of two molecules of water ?
(A) 5.98×10^{-23} g
(B) 36 g
(C) 18 g
(D) 36×10^{-23} g
70. CO₂ when passed in excess, in lime water turns colourless again because of
(A) Calcium carbonate
(B) Calcium bicarbonate
(C) Calcium chloride
(D) Copper carbonate



71. Which one of the following furnaces can produce the highest temperature ?
 (A) Blast furnace
 (B) Reverberatory furnace
 (C) Muffle furnace
 (D) Electric furnace
72. Which of the following statement is correct ?
 (A) Bauxite is an ore of aluminium
 (B) Magnetite is an ore of manganese
 (C) Haematite is an ore of mercury
 (D) Pyrites is an ore of phosphorus
73. Which of the following alkene will give the same product when treated with HBr in presence and absence of peroxide ?
 (A) Propene
 (B) Butene-2
 (C) 2-methylpropene
 (D) Butene-1
74. How many isomers does C_6H_{14} have ?
 (A) 3 (B) 6
 (C) 5 (D) 4
75. The nature of bonding in CCl_4 and CaH_2 is
 (A) Electrovalent in both CCl_4 and CaH_2
 (B) Covalent in CCl_4 and electrovalent in CaH_2
 (C) Electrovalent in CCl_4 and covalent in CaH_2
 (D) Covalent in both CCl_4 and CaH_2
76. NH_3 has a much higher boiling point than PH_3 , because
 (A) NH_3 has a largest molecular weight
 (B) NH_3 undergoes umbrella inversion
 (C) NH_3 forms hydrogen bond
 (D) NH_3 contains ionic whereas PH_3 contains covalent bonds
77. To which block do the elements with atomic number 56 belong ?
 (A) s (B) p
 (C) f (D) d
78. Which of the following atom has no neutron in its nucleus ?
 (A) Helium (B) Protium
 (C) Lithium (D) Tritium
79. Which among the following happens in an oxidation reaction ?
 (A) Electrons are lost
 (B) Electrons are gained
 (C) Protons are lost
 (D) Protons are gained
80. To which category of reactions does the following chemical reaction belong ?
 $NaCl(aq) + AgNO_3(aq) \rightarrow NaNO_3(aq) + AgCl(s)$
 (A) Redox reaction
 (B) Displacement reaction
 (C) Combination reaction
 (D) Double displacement reaction

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