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41. The most electropositive halogen is  
(A) F  
(B) Cl  
(C) Br  
(D) I
42. Elements having same valence electrons are placed in same group but there is an exception with  
(A) Ne (B) He  
(C) Ar (D) Xe
43. Which of the following gases will have the highest rate of diffusion ?  
(A) O<sub>2</sub> (B) CO<sub>2</sub>  
(C) NH<sub>3</sub> (D) N<sub>2</sub>
44. The potential energy of a molecule on the surface of liquid compared to one inside the liquid is  
(A) Zero  
(B) Smaller  
(C) The same  
(D) Greater
45. On increasing the concentration of reactants in a reversible reaction, the equilibrium constant will  
(A) Depend on the concentration  
(B) Increase  
(C) Unchanged  
(D) Decrease
46. Bee sting leaves an acid which causes pain and irritation. The acid released is  
(A) Tartaric acid  
(B) Citric acid  
(C) Ethanoic acid  
(D) Methanoic acid
47. What is called arenes ?  
(A) Aromatic hydrocarbons  
(B) Alkynes  
(C) Unsaturated hydrocarbons  
(D) Saturated hydrocarbons
48. Which of the following do not contain carbon-oxygen double bonds ?  
(A) Ketone (B) Esters  
(C) Acids (D) Ethers
49. Which among the following is the mass of two molecules of water ?  
(A)  $5.98 \times 10^{-23}$  g  
(B) 36 g  
(C) 18 g  
(D)  $36 \times 10^{-23}$  g
50. CO<sub>2</sub> when passed in excess, in lime water turns colourless again because of  
(A) Calcium carbonate  
(B) Calcium bicarbonate  
(C) Calcium chloride  
(D) Copper carbonate



51. Which one of the following furnaces can produce the highest temperature ?  
 (A) Blast furnace  
 (B) Reverberatory furnace  
 (C) Muffle furnace  
 (D) Electric furnace
52. Which of the following statement is correct ?  
 (A) Bauxite is an ore of aluminium  
 (B) Magnetite is an ore of manganese  
 (C) Haematite is an ore of mercury  
 (D) Pyrites is an ore of phosphorus
53. Which of the following alkene will give the same product when treated with HBr in presence and absence of peroxide ?  
 (A) Propene  
 (B) Butene-2  
 (C) 2-methylpropene  
 (D) Butene-1
54. How many isomers does  $C_6H_{14}$  have ?  
 (A) 3 (B) 6  
 (C) 5 (D) 4
55. The nature of bonding in  $CCl_4$  and  $CaH_2$  is  
 (A) Electrovalent in both  $CCl_4$  and  $CaH_2$   
 (B) Covalent in  $CCl_4$  and electrovalent in  $CaH_2$   
 (C) Electrovalent in  $CCl_4$  and covalent in  $CaH_2$   
 (D) Covalent in both  $CCl_4$  and  $CaH_2$
56.  $NH_3$  has a much higher boiling point than  $PH_3$ , because  
 (A)  $NH_3$  has a largest molecular weight  
 (B)  $NH_3$  undergoes umbrella inversion  
 (C)  $NH_3$  forms hydrogen bond  
 (D)  $NH_3$  contains ionic whereas  $PH_3$  contains covalent bonds
57. To which block do the elements with atomic number 56 belong ?  
 (A) s (B) p  
 (C) f (D) d
58. Which of the following atom has no neutron in its nucleus ?  
 (A) Helium (B) Protium  
 (C) Lithium (D) Tritium
59. Which among the following happens in an oxidation reaction ?  
 (A) Electrons are lost  
 (B) Electrons are gained  
 (C) Protons are lost  
 (D) Protons are gained
60. To which category of reactions does the following chemical reaction belong ?  
 $NaCl(aq) + AgNO_3(aq) \rightarrow NaNO_3(aq) + AgCl(s)$   
 (A) Redox reaction  
 (B) Displacement reaction  
 (C) Combination reaction  
 (D) Double displacement reaction

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