

COMBO - ଓଡ଼ିଶା
32,000- PYQ
ପୂର୍ବ ବର୍ଷ ପ୍ରଶ୍ନ
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41. Which one of the following oxides reacts with both acid and base ?
- (A) Aluminium Oxide
 - (B) Calcium Oxide
 - (C) Sodium Oxide
 - (D) Potassium Oxide
42. Which of the following element will become stable after losing an electron ?
- (A) Helium
 - (B) Iodine
 - (C) Sodium
 - (D) Oxygen
43. Ionic compounds consist of _____ charge ions.
- (A) oppositely
 - (B) equally
 - (C) negatively
 - (D) positively
44. What is the name of the graph that is drawn, when the temperature is kept constant ?
- (A) Isotherm
 - (B) Isochoric and isobar
 - (C) Isochoric
 - (D) Isobar
45. Surface Tension of a liquid is due to :
- (A) Gravitational force between molecules
 - (B) Electrical force between molecules
 - (C) Adhesive force between molecules
 - (D) Cohesive force between molecules
46. The neutralization reaction between an acid and a base is a type of :
- (A) Double displacement reaction
 - (B) Displacement reaction
 - (C) Addition reaction
 - (D) Decomposition reaction
47. The electronic configuration of an element M is 2, 8, 4 in modern periodic table. The element is placed in which group ?
- (A) 4th group
 - (B) 2nd group
 - (C) 14th group
 - (D) 18th group

48. The elements with atomic numbers 35, 53 and 85 are _____.
- (A) noble gases
(B) halides
(C) alkaline
(D) halogens
49. The element having tetra-atomic atomicity is _____.
- (A) Helium
(B) Nitrogen
(C) Phosphorus
(D) Chlorine
50. Reaction is said to be in equilibrium when :
- (A) the rate of transformation of reactants to products is equal to the rate of transformation of products to the reactants.
(B) 50% of the reactants are converted to products.
(C) the volume of the reactants is just equal to the volume of the products.
(D) the reaction is near completion and all the reactants are converted to products.
51. Which of the following are exothermic reactions ?
- (i) Reaction of water with quick lime
(ii) Dilution of an acid
(iii) Evaporation of water
(iv) Sublimation of Camphor (Crystals)
- (A) (i) and (ii)
(B) (ii) and (iii)
(C) (i) and (iv)
(D) (ii) and (iv)
52. Heating of Carbonate ores to remove carbon is called :
- (A) Roasting
(B) Calcination
(C) Smelting
(D) Fluxing
53. From the following which one is the example of chemical reaction ?
- (A) Grapes get fermented
(B) Breakdown of food
(C) Formation of curd
(D) All of the above

- character of elements on moving from left to right in a periodic table ?
- (A) Increases
(B) Decreases
(C) First increases then decreases
(D) First decreases then increases
55. The Conjugate acid of NH_2^- is :
- (A) NH_3
(B) NH_2OH
(C) NH_4^+
(D) N_2H_4
56. IUPAC name of $(\text{CH}_3)_2\text{CH}-\text{CH}=\text{CH}-\text{CH}_3$ is :
- (A) 2 methyl-3-pentene
(B) 4 methyl-2-pentene
(C) 2-isopropyl-1-propene
(D) 3-isopropyl-2-propene
57. The homolytic fission of a covalent bond liberates :
- (A) Carbonium ions
(B) Carbonions
(C) Free radicals
(D) Carbenes
- in nature ?
- (A) Iodine
(B) Sulphur
(C) Phosphorus
(D) Magnesium
59. Which of the following substances is not an aromatic compound ?
- (A) Benzene
(B) Naphthalene
(C) Anthracene
(D) Hexene
60. The cooking gas is mainly a mixture of the following two gases :
- (A) Methane and Ethane
(B) Ethane and Propane
(C) Propane and Butane
(D) Butane and Pentane



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