
26. 18g glucose is dissolved in 90g water the relative lowering in vapour pressure is equal to

- [A.](#) ? [B.](#) 5.1
[C.](#) 6 [D.](#) one/fifty one

Answer: Option D

27. Which of the following solution has the highest boiling point?

- [A.](#) 5.85% NaCl Solution [B.](#) 18.0 % glucose solution
[C.](#) 6.0 % urea solution [D.](#) All have same boiling point

Answer: Option A

28. Stronger is the oxidizing agent greater is the

- [A.](#) Oxidation potential [B.](#) Redox potential
[C.](#) e.m.f of cell [D.](#) standard reduction potential

Answer: Option D

29. Which of the following can be used in laptops?

- [A.](#) Silver oxide battery [B.](#) Fuel cell
[C.](#) Nickel cadmium cell [D.](#) Lead accumulator

Answer: Option C

30. Which is true about Zn-Cu galvanic cell?

- [A.](#) Reduction occurs at anode [B.](#) K⁺ ion transfer from salt bridge to left beaker of ZnSO₄
[C.](#) Oxidation occurs at cathode [D.](#) Anode is negatively charged

Answer: Option D

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