

31. The element that act as anode always have \_\_\_\_\_ position in electrochemical cell

- A. higher B. lower  
C. in middle D. no effect of position

**Answer:** Option A

32. In galvanic cell Zn acts as an anode so its value of standard reduction potential in comparison to coupled electrode would be

- A. greater B. lesser  
C. sum of reduction and oxidation potentials D. none

**Answer:** Option A

33. Greater value of standard reduction potential greater will be tendency

- A. to get oxidized B. to get reduced  
C. to accept electrons D. both b and c

**Answer:** Option D

34. Greater value of standard reduction potential smaller will be tendency

- A. to form positive ions B. to form negative ions  
C. gain electrons D. all are possible

**Answer:** Option A

35. Secondary cell is

- A. rechargeable B. non rechargeable  
C. electrolytic cell D. Daniel cell

**Answer:** Option A

